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Code	No.:133BG $\rightarrow$ ( ) $\rightarrow$ ( ) $\rightarrow$ ( ) $\rightarrow$ ( )
$\mathbf{J}$	AWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD
	B.Tech II Year I Semester Examinations, November/December - 2017
	METALLURGY AND MATERIALS SCIENCE
	(Common to ME, MCT, MSNT)
Time:	3 Hours Max. Marks: 75
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Note:	This question paper contains two parts A and B.
	Part A is compulsory which carries 25 marks. Answer all questions in Part A.
	Part B consists of 5 Units. Answer any one full question from each unit. Each question carries 10 marks and may have a, b, c as sub questions.
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	PART- A
	(25 Marks)
1.a)	Find the packing efficiency in HCP lattice. [2]
-( ) b)	Lattice parameter of a FCC crystal is 3.61A° calculate atomic density in (111), (110)
	and (100) planes. [3]
c)	What is the necessity of Alloying? [2]
d)	Distinguish between Intermetallic Compound and Electron compound. [3]
e)	What is congruent melting phase? [2]
f)	Define allotropy and give examples. [3]
(g)	What is ASTM-grain size number? What is its importance? [2]
——————————————————————————————————————	Distinguish between ordered and disordered solid solution. [3] ( )
i) - i) - i	What is coring and how it can be minimized? [2]
j)	What are the general requirements of a reinforcing phase? [3]
	PART-B
	(50 Marks)
2.a)	(50 Marks) What is an interstitial solid solution, name the five elements which commonly form
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<ul><li>→ (b)</li><li>3.a)</li></ul>	What is an interstitial solid solution, name the five elements which commonly form interstitial solid solutions? What is a grain size? What is a fine grained and coarse-grained material?  OR  What is crystal system and explain the Brevais lattices?
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